

Table 4. Explanation of descriptors used in Table 2.

#	Descriptor designation	Descriptor explanation
1	<i>RPCS</i>	Relative positive charged surface area
2	<i>DPSA</i> ⁽¹⁾	Difference in charged partial surface areas (Partial Positively charged Surface Area 1 - Partial Negatively charged Surface Area 1)
3	<i>FPSA</i> ⁽¹⁾	Fractional partial positively charged surface area (PPSA1/ Total molecular surface area)
4	<i>WPSA</i> ⁽¹⁾	Weighted partial positively charged surface area (PPSA1* Total molecular surface area /1000)
5	<i>WNSA</i> ⁽¹⁾	Weighted partial negatively charged surface area (PNSA1* Total molecular surface area /1000)
6	<i>DPSA</i> ⁽²⁾	Difference in charged partial surface areas (PPSA2-PNSA2)
7	<i>FNSA</i> ⁽²⁾	Fractional partial negatively charged surface area (PNSA-2/ Total molecular surface area)
8	<i>FPSA</i> ⁽²⁾	Fractional partial positively charged surface area (PPSA-2/ Total molecular surface area)
9	<i>WNSA</i> ⁽²⁾	Weighted partial negatively charged surface area (PNSA2* Total molecular surface area /1000)
10	<i>WPSA</i> ⁽²⁾	Weighted partial positively charged surface area (PPSA2* Total molecular surface area /1000)
11	<i>PPSA</i> ⁽³⁾	Atomic charge weighted partial positively charged surface area
12	<i>FNSA</i> ⁽³⁾	Fractional partial negatively charged surface area (PNSA3/ Total molecular surface area)
13	<i>FPSA</i> ⁽³⁾	Fractional partial positively charged surface area (PPSA3/ Total molecular surface area)
14	<i>WNSA</i> ⁽³⁾	Weighted partial negatively charged surface area (PNSA3* Total molecular surface area /1000)
15	<i>WPSA</i> ⁽³⁾	Weighted partial positively charged surface area (PPSA3* Total molecular surface area /1000)
16	⁰ <i>CIC</i>	Complementary Information content (order 0)
17	⁰ \overline{CIC}	Average Complementary Information content (order 0)

18	${}^0\overline{BIC}$	Average Bonding Information content (order 0)
19	${}^0\overline{SIC}$	Average Structural Information content (order 0)
20	1CIC	Complementary Information content (order 1)
21	${}^1\overline{CIC}$	Average Complementary Information content (order 1)
22	${}^1\overline{BIC}$	Average Bonding Information content (order 1)
23	${}^2\overline{IC}$	Average Information content (order 2)
24	2SIC	Structural Information content (order 2)
25	${}^2\overline{SIC}$	Average Structural Information content (order 2)
26	q_A^{\min}	Minimum partial charge for all atom types
27	q_A^{\max}	Maximum partial charge for all atom types
28	q_C^{\max}	Maximum net atomic charge (typed) for atom <i>C</i>
29	q_C^{\min}	Minimum net atomic charge (typed) for atom <i>C</i>
30	q_{net}^{\min}	Minimum net atomic charge
31	V_C^{\min}	Minimum valency for atom <i>C</i>
32	V_H^{\min}	Minimum valency for atom <i>H</i>
33	\overline{V}_H	Average valency for atom <i>H</i>
34	P_H^{\min}	Minimum bond order for atom <i>H</i>
35	\overline{P}_C	Average bond order for atom <i>C</i>
36	R_C^{\min}	Minimum one-electron reactivity index for atom <i>C</i>
37	E_{tot}	Total molecular electrostatic interaction
38	$E_{exc}^{\min}(H-C)$	Minimum exchange energy for bond <i>H-C</i>
39	$E_{state}^{\min}(H)$	Minimum atomic state energy for atom <i>H</i>
40	$N^{occ.el.lev}$	Number of occupied electronic levels
41	$N^{occ.el.lev} / N_A$	Number of occupied electronic levels / number of atoms

42	n_A^{\max}	Maximum atomic orbital electronic population
43	<i>HOMO</i>	Energy of the highest occupied molecular orbital
44	${}^0\chi$	Randic index (order 0)
45	${}^0\chi^v$	Kier&Hall index (order 0)
46	${}^1\chi$	Randic index (order 1)
47	${}^1\chi^v$	Kier&Hall index (order 1)
48	${}^3\chi$	Randic index (order 3)
49	<i>J</i>	Balaban index
50	<i>W</i>	Wiener index
51	Φ	Kier flexibility index
52	N^{HA}	Count of <i>H</i> -acceptor sites
53	N^{HD}	Count of <i>H</i> -donor sites
54	$(HA, HD)^{\min}$	Minimum value of N^{HA} and N^{HD}
55	$(HA, HD)^{\max/\min}$	Maximum/minimum ratio between N^{HA} and N^{HD} values
56	<i>HDCA</i>	Hydrogen donors charged surface area
57	<i>HASA</i>	Hydrogen acceptors surface area
58	<i>FHACA</i>	Fractional hydrogen acceptor charged surface area (HACA/ Total molecular surface area)
59	<i>FHASA</i>	Fractional hydrogen acceptors surface area (HASA/ Total molecular surface area)
60	<i>FHBSA</i>	Fractional hydrogen bonding surface area (HBSA/ Total molecular surface area)
61	${}^{HA}HDCA^{(1)}$	Hydrogen acceptor dependent Hydrogen donors charged surface area -1
62	$HACA^{(1)}$	Hydrogen acceptors charged surface area - 1
63	$HACA^{(2)}$	Hydrogen acceptors charged surface area -2
64	${}^{HA}PSA^{(2)}$	New hydrogen acceptors partial surface area
65	${}^{HA}HDCA^{(2)}$	Hydrogen acceptor dependent HDCA-2

66	$^{HA}HDSA^{(2)}$	Hydrogen acceptor dependent Hydrogen donors surface area-2
67	$^{HD}CPSA^{(2)}$	New hydrogen donors charged partial surface area
68	$^{HA}FPSA^{(2)}$	New hydrogen acceptors fractional partial surface area
69	$^{HD}FPSA^{(2)}$	New hydrogen donors fractional partial surface area
70	$^{HA}FCPSA^{(2)}$	New hydrogen acceptors fractional charged partial surface area
71	$^{HD}FCPSA^{(2)}$	New hydrogen donors fractional charged partial surface area
72	S_{ZX}	Shadow plane ZX
73	S_{YZ}	Shadow plane YZ
74	S_{XY} / R_{XY}	XY Shadow / XY Rectangle
75	S_{YZ} / R_{YZ}	YZ Shadow / YZ Rectangle
76	V_M / V_{XYZ}	Molecular volume / XYZ Box
77	G_b	Gravitation index (all bonds)
78	G_p	Gravitation index (all atoms' pairs)
79	T^E	Topographic electronic index (all atoms' pairs)
80	T_b^E	Topographic electronic index (all bonds)
81	α	Alfa polarizability
82	μ	Total dipole of the molecule
83	μ_{hyb}	Total hybridization component of the molecular dipole
84	μ_{pch}	Total point-charge component of the molecular dipole
85	S_M	Total molecular surface area
86	ΔH_f^0	Thermodynamic heat of formation of the molecule at 300K
87	$\Delta H_f^0 / N_A$	Thermodynamic heat of formation of the molecule at 300K / number of atoms
88	ΔS_{tot}	Total entropy (300K)
89	$\Delta S_{vib} / N_A$	Vibration entropy (300K) /number of atoms

90	c_p^{int} / N_A	Internal heat capacity (300K) / number of atoms
100	ν_{TD}^h	Highest normal mode vibration transition dipole
101	P_f^2	Polarity parameter / (square distance) ²